

UNITED STATES  
DEPARTMENT OF  
COMMERCE  
PUBLICATION



# NBS TECHNICAL NOTE 608

U. S.  
DEPARTMENT  
OF  
COMMERCE

National  
Bureau  
of  
Standards

QC  
100  
.U5753  
No.608  
1971  
Copy 2.

## Steam-Water, Critical Flow in a Venturi

## NATIONAL BUREAU OF STANDARDS

The National Bureau of Standards<sup>1</sup> was established by an act of Congress March 3, 1901. The Bureau's overall goal is to strengthen and advance the Nation's science and technology and facilitate their effective application for public benefit. To this end, the Bureau conducts research and provides: (1) a basis for the Nation's physical measurement system, (2) scientific and technological services for industry and government, (3) a technical basis for equity in trade, and (4) technical services to promote public safety. The Bureau consists of the Institute for Basic Standards, the Institute for Materials Research, the Institute for Applied Technology, the Center for Computer Sciences and Technology, and the Office for Information Programs.

**THE INSTITUTE FOR BASIC STANDARDS** provides the central basis within the United States of a complete and consistent system of physical measurement; coordinates that system with measurement systems of other nations; and furnishes essential services leading to accurate and uniform physical measurements throughout the Nation's scientific community, industry, and commerce. The Institute consists of a Center for Radiation Research, an Office of Measurement Services and the following divisions:

Applied Mathematics—Electricity—Heat—Mechanics—Optical Physics—Linac Radiation<sup>2</sup>—Nuclear Radiation<sup>2</sup>—Applied Radiation<sup>2</sup>—Quantum Electronics<sup>3</sup>—Electromagnetics<sup>3</sup>—Time and Frequency<sup>3</sup>—Laboratory Astrophysics<sup>3</sup>—Cryogenics<sup>3</sup>.

**THE INSTITUTE FOR MATERIALS RESEARCH** conducts materials research leading to improved methods of measurement, standards, and data on the properties of well-characterized materials needed by industry, commerce, educational institutions, and Government; provides advisory and research services to other Government agencies; and develops, produces, and distributes standard reference materials. The Institute consists of the Office of Standard Reference Materials and the following divisions:

Analytical Chemistry—Polymers—Metallurgy—Inorganic Materials—Reactor Radiation—Physical Chemistry.

**THE INSTITUTE FOR APPLIED TECHNOLOGY** provides technical services to promote the use of available technology and to facilitate technological innovation in industry and Government; cooperates with public and private organizations leading to the development of technological standards (including mandatory safety standards), codes and methods of test; and provides technical advice and services to Government agencies upon request. The Institute also monitors NBS engineering standards activities and provides liaison between NBS and national and international engineering standards bodies. The Institute consists of the following technical divisions and offices:

Engineering Standards Services—Weights and Measures—Flammable Fabrics—Invention and Innovation—Vehicle Systems Research—Product Evaluation Technology—Building Research—Electronic Technology—Technical Analysis—Measurement Engineering.

**THE CENTER FOR COMPUTER SCIENCES AND TECHNOLOGY** conducts research and provides technical services designed to aid Government agencies in improving cost effectiveness in the conduct of their programs through the selection, acquisition, and effective utilization of automatic data processing equipment; and serves as the principal focus within the executive branch for the development of Federal standards for automatic data processing equipment, techniques, and computer languages. The Center consists of the following offices and divisions:

Information Processing Standards—Computer Information—Computer Services—Systems Development—Information Processing Technology.

**THE OFFICE FOR INFORMATION PROGRAMS** promotes optimum dissemination and accessibility of scientific information generated within NBS and other agencies of the Federal Government; promotes the development of the National Standard Reference Data System and a system of information analysis centers dealing with the broader aspects of the National Measurement System; provides appropriate services to ensure that the NBS staff has optimum accessibility to the scientific information of the world, and directs the public information activities of the Bureau. The Office consists of the following organizational units:

Office of Standard Reference Data—Office of Technical Information and Publications—Library—Office of Public Information—Office of International Relations.

<sup>1</sup> Headquarters and Laboratories at Gaithersburg, Maryland, unless otherwise noted; mailing address Washington, D. C. 20234.

<sup>2</sup> Part of the Center for Radiation Research.

<sup>3</sup> Located at Boulder, Colorado 80302.

UNITED STATES DEPARTMENT OF COMMERCE

Maurice H. Stans, Secretary

U.S. NATIONAL BUREAU OF STANDARDS • Lewis M. Branscomb, Director



# TECHNICAL NOTE 608

ISSUED JULY 1971

Nat. Bur. Stand. (U.S.), Tech. Note 608, 25 pages (July 1971)

CODEN: NBTNA

## Steam-Water, Critical Flow in a Venturi

R. V. Smith

Cryogenics Division  
Institute for Basic Standards  
National Bureau of Standards  
Boulder, Colorado 80302



NBS Technical Notes are designed to supplement the Bureau's regular publications program. They provide a means for making available scientific data that are of transient or limited interest. Technical Notes may be listed or referred to in the open literature.



## Contents

	Page
1. Introduction . . . . .	1
2. Experimental . . . . .	4
3. Qualitative Observations . . . . .	4
4. Analytical Procedure . . . . .	8
5. Results and Discussion . . . . .	14
6. Conclusions . . . . .	16
7. Acknowledgments . . . . .	17
8. References . . . . .	18
9. Nomenclature . . . . .	20

## List of Figures

	Page
Figure 1. Flow circuit of the steam-water loop. . . .	5
Figure 2. Pressure versus distance along the venturi showing the effect of raising the downstream pressure . . . .	6
Figure 3. Pressure versus distance along the venturi. Curves are from the analytical program and points from the experimental program. . . . .	12
Figure 4. Pressure versus distance along the venturi. Curves are from the analytical program and points from the experimental program. . . . .	13

# STEAM-WATER, CRITICAL FLOW IN A VENTURI

R. V. Smith

This paper is the second part of an analytical and experimental investigation, in which the primary object was to test the hypothesis that the flow of the gas phase controls critical and near critical two-phase flow for cases where the gas and liquid flow essentially in separate streams. In the first part of the investigation, a two-component system (air-water) was used. The results presented here substantiate the hypothesis. The analytical results also indicate the use of one dimensional flow equations with reasonably accurate estimates for droplet size and for the drag and heat transfer coefficients (without consideration of mass transfer--vaporization or condensation) describe critical and near-critical flow reasonably well. This indicates that mass transfer may be a secondary effect for these flow conditions.

Key words: Critical flow; pressure profile; steam; venturi; water.

## 1. Introduction

This study was the second part of a two-phase critical flow program. In the first part, (Smith 1968, 1971) a two-component, two-phase system (water and air) was employed, and the experimental program was quite extensive. This second part was an extension of that work, and was conducted primarily to compare two and one component behavior. The analytical program was essentially the same as that used for the two component system. The hypothesis to be tested for both studies was that the gas behavior primarily controls the critical (choking) two-phase flow for higher qualities where essentially all the

gas flows in separated and continuous streams. A more detailed description of previous work and objectives of this program may be found in Smith (1968, 1971).

This study differs from many of the previous works in that it follows the flow from an upstream point and it defines critical flow as a state that occurs when the gas, rather than the mixture, reaches critical conditions. In the largest number of studies the primary interest has been in critical flow at the geometric point where critical flow occurs. This restriction of the study is useful, but it imposes several disadvantages. One is that the fluid conditions at the critical point are difficult to determine. The pressure gradient in that region is very steep and experimental measurements from conventional wall taps in a converging or straight section must be extrapolated to the point of critical flow. Also, in one component flow the quality or gas-liquid distribution cannot be directly measured and must be calculated by various equilibrium or phase-rate-of-change assumptions. In almost all works at the critical point, thermal equilibrium has been assumed. The first phase of this study and other works, such as Fauske (1965) and Carofano and McManus (1968), indicate that this is not a valid assumption.

Further, the velocity of each phase cannot be determined accurately because experimental void (gas-volume fraction) measurements are so difficult to obtain that such velocity data must be evaluated with the use of uncertain quality data. These disadvantages make data for the point of critical flow of limited use, for example, for designers of flow systems.

Nevertheless, work at the point of critical flow has resulted in fairly reliable expressions for the critical flow rate, for example, Fauske (1961), Moody (1965), Zivi (1964) and Cruver and Moulton (1966).



These works, although they have the same general range of reliability of prediction, vary widely in assumptions regarding equilibrium and the gas and liquid velocities. The assumptions made for the gas-liquid velocity ratio are, in almost all cases, substantially higher than experimental measurements reported by Fauske (1965) and Klingbiel (1964).

In the program described here, an attempt was made to minimize and resolve some of these difficulties by studying the flow from points upstream where fluid conditions are known more reliably. Thus, estimates regarding mass, momentum and energy transport can be made throughout the region of rapid acceleration to critical flow. These estimates can then be checked by comparison of analytical and experimental pressure profiles. This procedure of analysis from an upstream point is also followed in a more recent work by Henry and Fauske (1971) and that approach together with the interface analysis procedure reported here is very similar to that reported by Carofano and McManus (1969). This work and that of Carofano and McManus were developed independently at about the same time.

The second difference between this study and most previous works lies in the analytical conditions prescribed for critical flow. For the most part, in previous work the fluid has been treated as a mixture and critical flow has been related to mixture properties. In this treatment, dealing with cases where the flow is essentially separated it has been assumed that critical flow occurs when the gas reaches critical flow. This assumption for critical condition is the same as earlier works using the "vapor choking" model at the point of critical flow, for example, Ryley (1961).

## 2. Experimental Procedure

The experimental apparatus was similar to, but smaller than, that used in the first part and is shown in figure 1. The two-phase mixture comes from the heater and flows through the annulus of the venturi. The rack and pinion permitted the venturi core to be moved axially, which allowed axial pressure measurements to be closely spaced near the throat where the gradient was very steep. The support for the venturi core was rigid with concentricity thoroughly checked before installation. Any possible variations could be observed visually and by examination of motion pictures and still photographs taken during the runs. The quantity of liquid film flow at the tube wall was measured by extracting the film. The ratio of liquid at the wall to total liquid flow ( $m_{ff}/m_f$ ) is shown in figures 3 and 4. The mixture quality at the venturi entrance was determined by calculations which took into consideration the energy input and in which thermal equilibrium was assumed. Critical flow at the throat was verified by the presence of a shock wave in the divergent portion of the venturi. If the shock were very near the throat, the readings in that region could have been disturbed, so only data from runs where the shock was greater than 0.05 inches downstream of the throat are reported. Liquid flow was measured by calibrated rotameters between the condenser and the heaters. Pressure was measured at the entrance and near the throat by transducers. Estimated  $3\sigma$  error in the flowmeter reading was 3% and in the pressure readings was 4%.

## 3. Qualitative Observations

As in the case for air-water flow (Smith, 1968, 1971), there were some qualitative observations which support the hypothesis of gas-controlled flow. As shown in figure 2, after critical flow was estab-

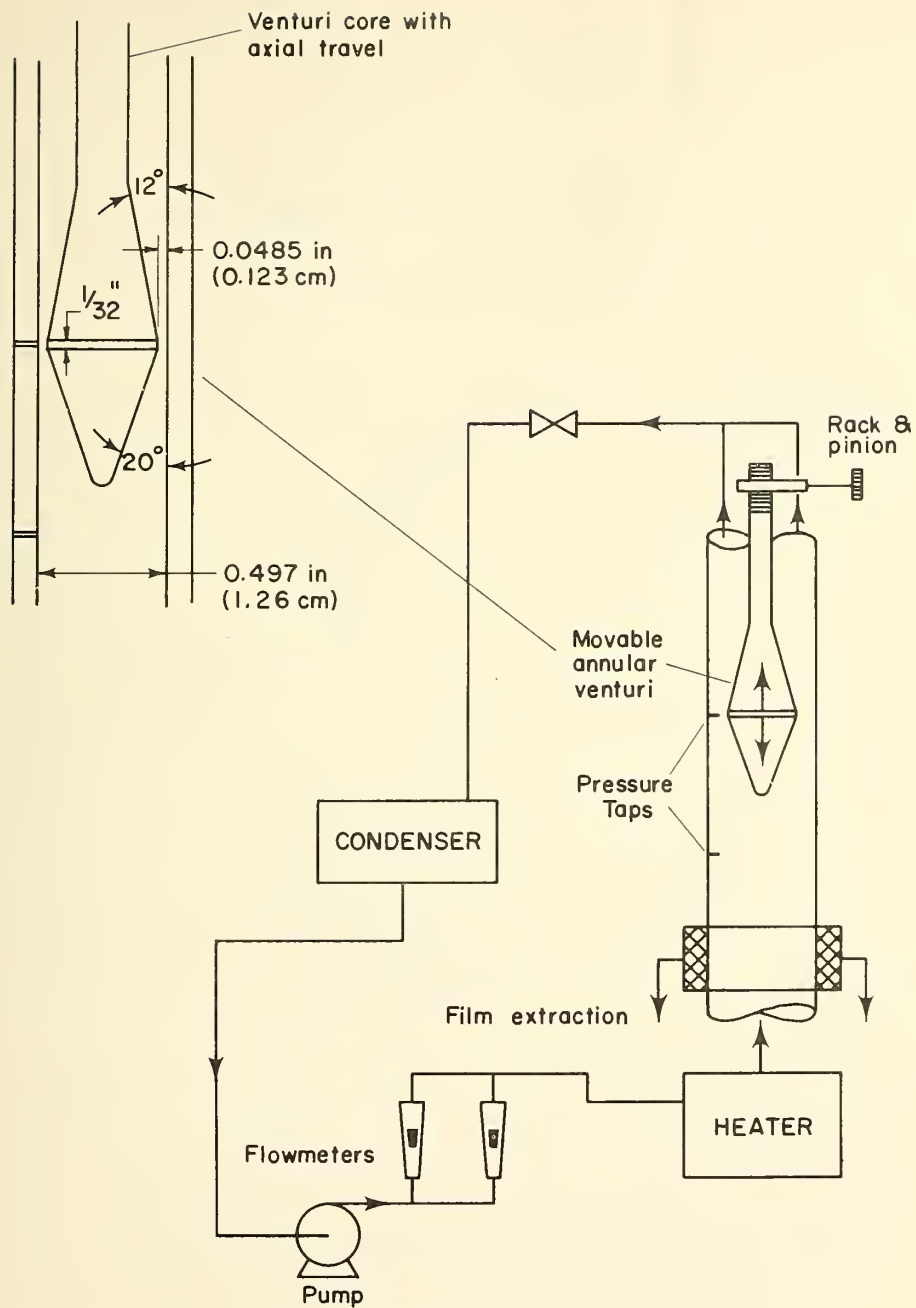


Figure 1. Flow circuit of the steam-water loop

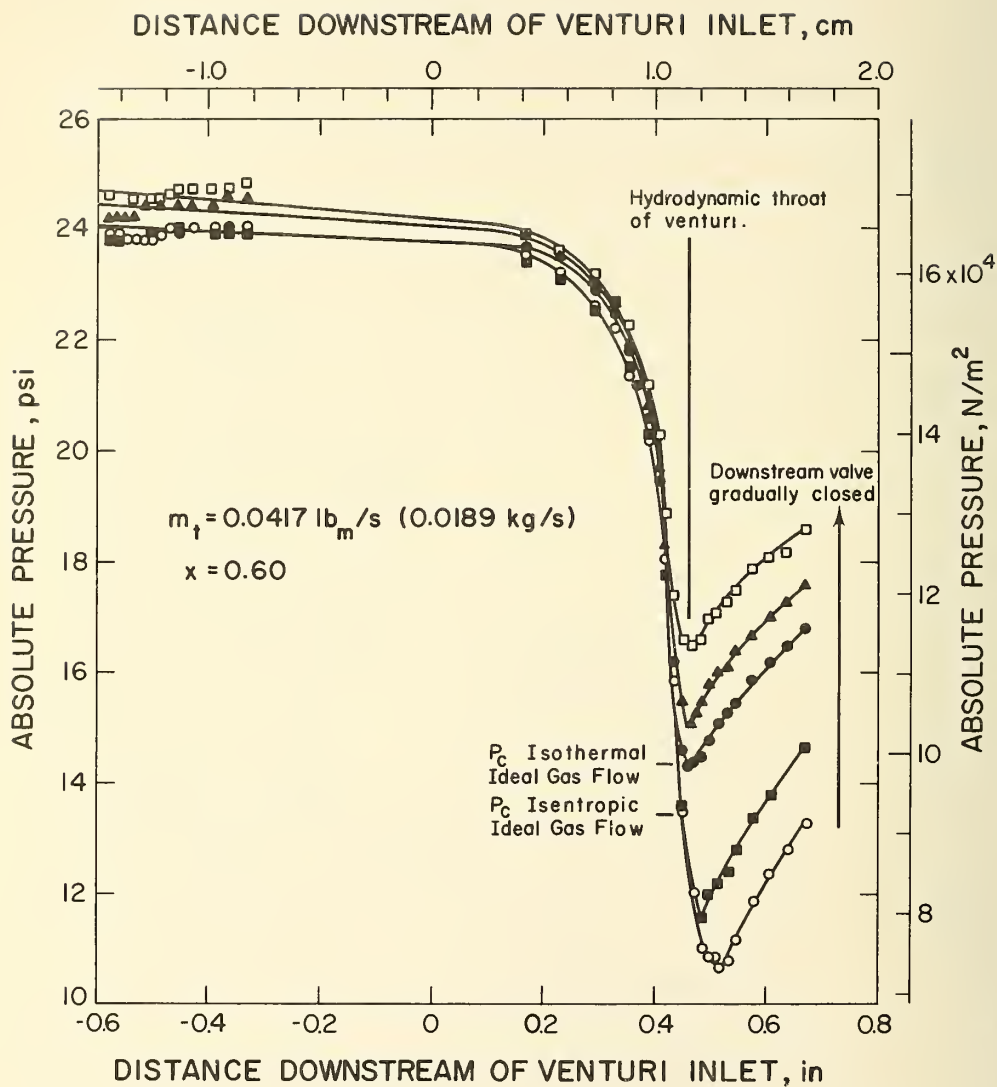


Figure 2. Pressure versus distance along the venturi showing the effect of raising the downstream pressure

lished, the downstream pressure was raised from its minimal value and this moved the shock wave toward the throat. When the shock reached the throat, the pressure was approximately that which would be expected for ideal steam flow alone.

Also, when the standing shock wave reached the throat, the changes in the downstream pressure were transmitted upstream. If the gas were not controlling the critical flow behavior, one might expect pressure signals to be transmitted upstream at different velocities. For example, some pressure signals might be transmitted through a continuous liquid phase at a high velocity and have some influence on the upstream pressure throughout the range of downstream pressure changes. Since the upstream pressure was not noticeably changed until the shock reached the throat, it would appear that the vapor flow characteristics were controlling.

The accuracy of the experimental critical pressure ratios ( $p_c/p_o$ ) was estimated to be within  $\pm 0.03$ . This was primarily a result of the uncertainty in the location of the hydrodynamic venturi throat. The location was fixed by studying runs where the back pressure was raised to produce subcritical flow, as shown in figure 2. The hydrodynamic throat was identified as the minimum pressure point for the sub-critical runs, where downstream pressure changes changed the upstream pressure. Smith (1968) showed that with a similar venturi the geometric throat, the hydrodynamic throat as defined here and the throat indicated by pressure measurements with critical, all gas flow all fell within an axial range of 0.020 inches. It was concluded that the relationship would be approximately the same for the case reported here. As studies following this general procedure are continued, further detailed examination of the critical flow at the throat is indicated. These critical pressure ratios varied from 0.5 to 0.6, considering the

experimental uncertainty, in the general range indicated for isentropic to isothermal flow of the vapor alone, which is 0.55 to 0.61. Of course, vaporization or condensation and momentum transport would change the value of  $p_c/p_o$  from that for ideal gas alone. It would appear, however, that the general agreement, between the experimental data and the expected behavior for the ideal gas flow indicates that the characteristics of the gas flow was the primary factor in controlling the critical flow behavior for the conditions reported. Critical pressure ratios in this range were also reported by Carofano and McManus (1969) for a steam-water system with similar flow conditions.

#### 4. Analytical Procedure

The analytical model was the same as that used for the air-water portion of the program. Thus, the rates of vaporization or condensation were considered zero or negligible. It was recognized that the vaporization and condensation effects could be significant. However, it seemed useful to compare data processed using this "frozen flow" model with the experimental data.

Briefly, the analytical model consisted of a liquid film flowing at the tube wall and a gas core containing entrained liquid droplets. As mentioned previously, the distribution of liquid between film and droplets was experimentally determined by extracting and measuring the rate of the liquid film flow. Assuming one dimensional flow, working expressions were derived from the conservation equations. Frictional pressure drop, due to forces at the tube wall, was considered negligible compared to the momentum pressure drop in equations (1,2). Potential energy terms such as changes in elevation, and pressure changes were neglected in the energy expression (6).

The momentum equation for the mixture was written as:

$$- dp/d\ell = \frac{m_g (du_g/d\ell) + m_{fe} (du_{fe}/d\ell) + m_{ff} (du_{ff}/d\ell)}{A_t} . \quad (1)$$

Fluid velocities from momentum and continuity equations were:

$$\text{for gas velocity } du_g/d\ell = \frac{-(dA_g/d\ell) u_g}{A_g [1 - u_g^2/(dp/d\rho_g)]} , \quad (2)$$

for liquid velocity

$$du_{ff}/d\ell = \frac{-A_{ff}(dp/d\ell) - \rho_f A_{ff} g + 2\pi r_{fg} \rho_g [(u_g - u_{ff})^2/2]}{m_{ff}} , \quad (3)$$

and for liquid droplet velocity

$$du_{fe}/d\ell = \frac{\rho_g \left[ \frac{(u_g - u_{fe})^2}{2} \right] A_d C_{dr}}{\frac{4}{3} \pi R_d^3 \rho_f u_{fe}} . \quad (4)$$

Energy transport and fluid temperatures from energy and heat transfer equations were:

for interface energy transport

$$q = h_c A_{fg} (T_g - T_f) , \quad (5)$$



for gas temperature

$$dT_g/d\ell = \frac{u_g (du_g/d\ell) + (m_{fe}/m_g) u_{fe} (du_{fe}/d\ell)}{c_{pg}} + \frac{(m_{ff}/m_g) u_{ff} (du_{ff}/d\ell) + 1/m_g (dq/d\ell)}{c_{pg}}, \quad (6)$$

and for liquid temperature

$$dT_f/d\ell = \frac{dq/d\ell}{m_f c_f}. \quad (7)$$

Gas density from the ideal gas equation of state was

$$d\rho_g/d\ell = \frac{dp/d\ell - R\rho_g (dT_g/d\ell)}{RT_g}. \quad (8)$$

Effective gas flow area, empirically adjusted to account for waves and droplet distribution, was:

$$A_{gE} = A_t - A_{f \text{ str}} - C_l A_{f \text{ str}} l. \quad (9)$$

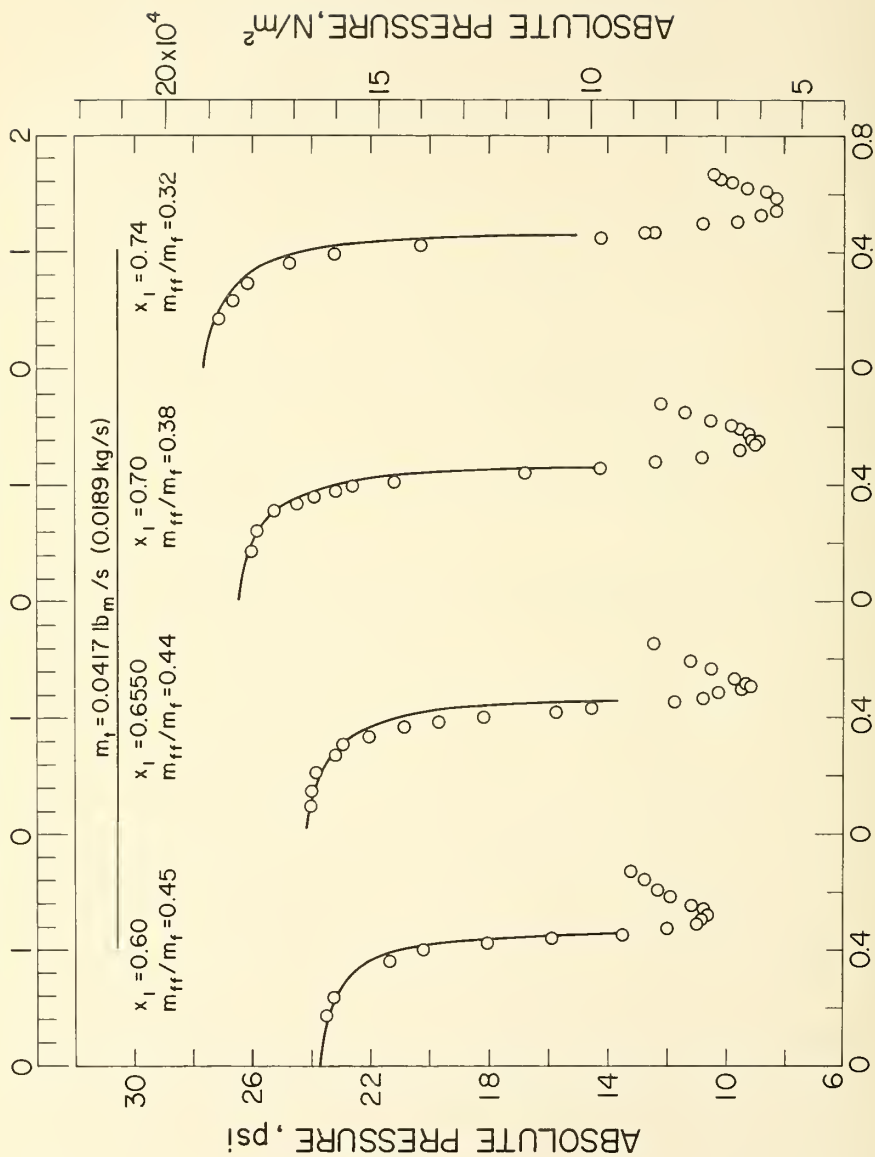
After considerable study, the droplet diameter, the values or expressions for drag coefficients, and the interface heat transfer coefficients were determined. These were selected from data reported in the literature for similar cases and, from among that group, were primarily the values or expressions which best fitted the experimental, critical pressure-ratio data for air-water flow. It was found that variations in the assigned values of these terms by factors of two to four produced computed critical pressure ratio changes no greater than five percent, except in the case of droplet radii, where the variations were



as high as ten percent. Therefore, the final results were not particularly sensitive to the assigned values for these terms. Droplet radii,  $R_d$ , data were from Wicks et al. (1966) and Ryley (1961). Droplet drag coefficients,  $C_{dr}$ , involved three studies, Ingebo (1956), Lappel (1950), and Rabin et al. (1960), with the last of these chosen as best describing the data. The liquid-gas interface drag coefficient,  $f_{fg}$ , was set at the equivalent fluid-solid interface value for complete turbulence with rough tubes, using conventional friction factor curves. The convective heat transfer coefficient,  $h_{ce}$ , for the droplets was of the Ranz and Marshall (1952) type for laminar flow. The interface heat transfer coefficient for the liquid film,  $h_{cf}$ , was the value for a fluid-solid system in turbulent flow, Dittus and Boelter (1930), and this value was partially confirmed by temperature measurements of the liquid film.

The effective gas flow area in (9) was determined by first subtracting a computed liquid flow area, which would occur for the liquid flowing in a smooth stream, and then subtracting the final term in the equation, which is a blockage factor, to account for the reduced gas flow area caused by waves and by droplet distributions. The value of  $C_1$ , which is a multiplier for the venturi-inlet, stream liquid flow areas, was empirically determined so that the analytical program would achieve critical flow at the venturi throat. Although the value of this coefficient did empirically fix the location of critical flow for the analytical program by adjusting the effective gas flow area, it did not influence the relationship in equations (1-8), which determined the pressure distribution that was used to compare analytical and experimental results. The value of  $C_1$  varied between 1.5 and 3.0 for the range of experimental data reported. Although the higher values of  $C_1$  may seem large, one must remember that this is an adjustment

DISTANCE DOWNSTREAM OF VENTURI INLET, cm



DISTANCE DOWNSTREAM OF VENTURI INLET, in

Figure 3. Pressure versus distance along the venturi.  
Curves are from the analytical program and points from the experimental program.

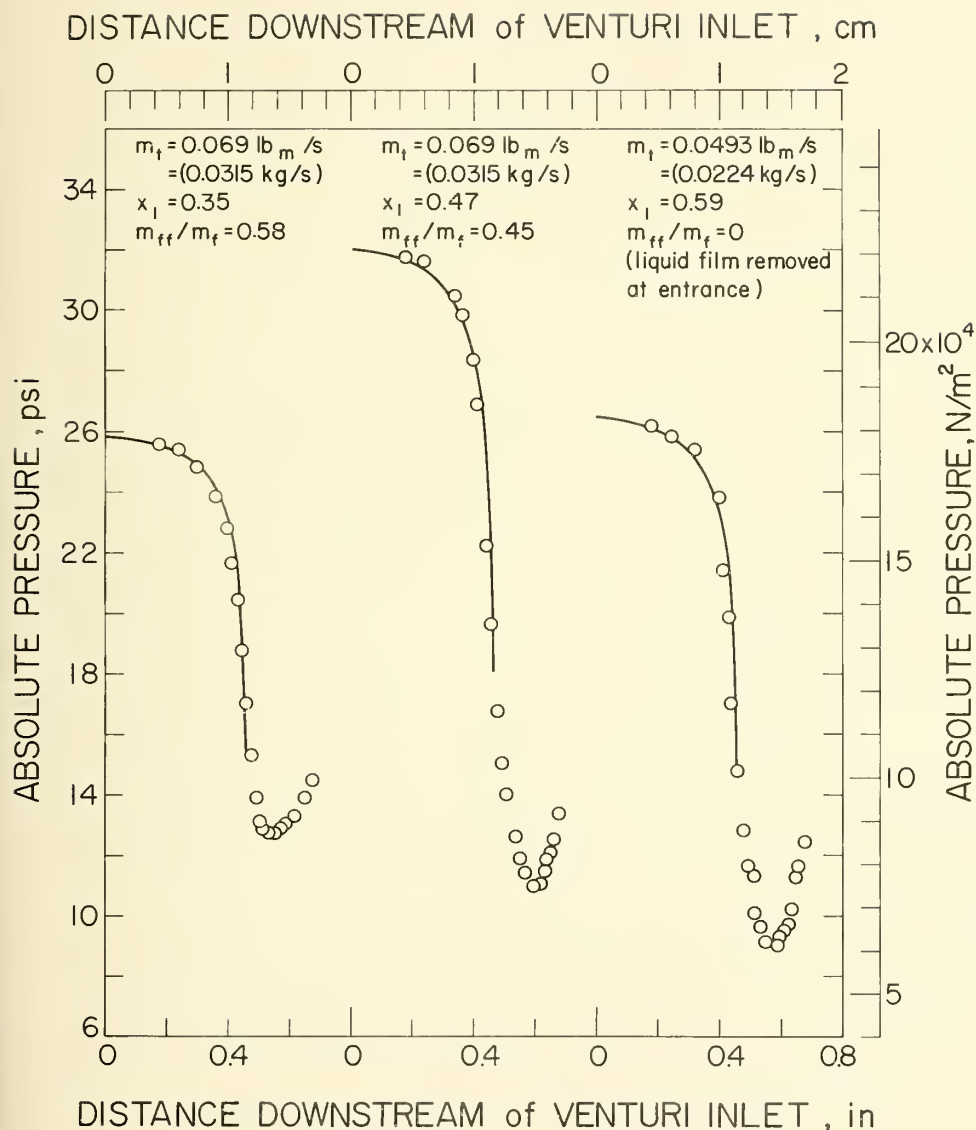


Figure 4. Pressure versus distance along the venturi. Curves are from the analytical program and points from the experimental program.

from a liquid-flow model which is far from realistic. The multiplier  $C_1$  is applied to the stream flow area, which at these high qualities, is very small compared to the total area. Also, estimates of wave behavior and droplet distribution indicate that these blockage values are in a reasonable range, (Smith, 1968).

The computational procedure was to numerically integrate the flow equations from the entrance point of the venturi to the point of critical flow, using the Runge-Kutta (1962) method. The axial length of step between computation points was determined by,

$$\Delta \ell_{\text{step}} = C_2 [1 - \sqrt{\ell} / C_3]. \quad (10)$$

The constants  $C_2$  and  $C_3$  were adjusted to produce the required accuracy in the final numerical results. This demanded that  $C_3$  be approximately the value of the length of the converging section in the venturi, to insure a very short step length near the throat where the pressure gradient is very steep. The point of critical flow was detected in the numerical system by examining the behavior of (2). If the gas flow is primarily controlling, the mixture flow will be critical when the gas flow reaches the critical condition. For that case, the denominator of (2) becomes zero. The maximum uncertainty for this computational procedure was estimated at plus or minus 0.01, for example, for the critical pressure ratio.

## 5. Results and Discussion

The analytical and experimental results are shown in figures 3 and 4. These data are for different total mass flow rates, all at relatively high quality.

Both experimental and analytical data show a typical, gas-flow profile. The agreement between the analytical and experimental data is quite good, especially considering the "frozen" flow (no vaporization or condensation) model used. The agreement appears to be somewhat poorer than that for the air-water data. The steeper slope (tendency for the experimental pressures to be lower) which appears to prevail for the experimental data may indicate the influence of the vaporization or condensation omitted in the analytical procedure. One should note, however, that in these figures the slope of the pressure curve is very steep and this tends to make the agreement between the experimental and analytical data appear somewhat better than it actually is.

In comparing these data with air-water data from Smith (1968, 1971), the difference in the agreement between the experimental and analytical profiles appears significant but not great. This rather strongly indicates that the gas behavior exerts primary control. It may also indicate that the influence of condensation and vaporization, while noticeable, is of a secondary nature for these flow conditions.

Carofano and McManus (1969), using very similar analytical and experimental procedures, found what appears to be about the same level of agreement with the pressure profiles. In their analytical model, they took into consideration condensation and vaporization rates while employing a fog flow (liquid totally entrained) model. Rough comparison of critical mass flow rates on runs with similar entrance fluid conditions showed agreement at least within  $\pm 5\%$  between the experimental critical flow rate data reported here and that of Carofano and McManus (1969).

Calculation of the critical mass flow rate,  $G$ , using the equation of Cruver and Moulton (1966),

$$\frac{m_t}{A} = \left[ \frac{-(u_g/u_f)}{C_{A1} (dv_g/dp)_s + C_{A1} (dx/dp)_s + C_{A3} (dv_f/dp)_s} \right]^{1/2}, \quad (11)$$

where

$$C_{A1} = x \left[ 1 + \frac{[(u_g/u_f) - 1]}{3 u_g/u_f} [x(3u_g/u_f + 1) - 1] \right],$$

and

$$C_{A2} = u_g \left[ 1 + 2x (u_g/u_f - 1) \right] + u_f (u_g/u_f) \left[ (1 - 2x) (u_g/u_f - 1) \right],$$

and

$$C_{A3} = (u_g/u_f) (1 - x) \left[ 1 + x (u_g/u_f - 1) (1 + u_g/3u_f) \right],$$

with  $(u_g/u_f) = (\rho_f/\rho_g)^{1/3}$ , predicted mass flow rates about double of those reported here. This is not unexpected, as in that type of equation, thermal equilibrium to the throat is assumed. For most of the runs reported here, this assumption would predict superheated steam at the throat. Therefore, it can only be concluded that the Cruver and Moulton type of equation will substantially overpredict the flow rate if the entrance quality is used; since liquid was always observed at the throat, the equilibrium assumption is known to be incorrect. Cruver and Moulton (1966) and Fauske (1965) have suggested that this type of equation may use a slip ratio higher than the actual ratio to compensate for the error in the equilibrium assumption. These data would tend to support that suggestion.

## 6. Conclusions

This work, following a more comprehensive air-water study, further substantiates the concept that gas flow behavior is the primary factor in controlling critical, two-phase flow for higher void or

quality flow conditions where the phases flow separately. The analytical model for one-dimensional flow, with estimates for interface phenomena of momentum and energy transport, but neglecting mass transport (vaporization or condensation), showed good agreement with the pressure profiles just upstream of and at the critical point. The critical pressure ratios were in general agreement with those expected for ideal gas flow.

## 7. Acknowledgments

The experimental work for this program was carried out during a period when the author was attached to UKARE, Harwell England. Mr. L. B. Cousins and Dr. G. F. Hewitt made substantial contributions to the program.



## 8. References

- Carofano, G. C. and McManus, H. N. (1969), "An analytical and experimental study of the flow of air-water and steam water mixtures in a converging-diverging nozzle", Progress in Heat and Mass Transfer, Vol. 2 (Pergamon Press).
- Cruver, J. E., and R. W. Moulton, (1966), "Metastable critical flow of steam-water mixtures," AIChE, Fundamentals of Fluid Mechanics Symposium, Detroit.
- Dittus, F. W., and L. M. K. Boelter, (1930), Univ. of Calif. Pubs. Eng. 2, 443.
- Fauske, H. K., (1965), "Two-phase two-and-one-component critical flow," Proc. Symp. on two-phase flow, U. of Exeter, England.
- Fauske, H., (1961), "Critical, two-phase, steam-water flows," Proc. 1961 Heat Transfer and Fluid Mech., Inst. Stanford Univ. Press, 79.
- Henry, R. E. and Fauske, H. K. (1971), "The two-phase critical flow of one-component mixtures in nozzles, orifices and short tubes," ASME paper No. 70-WA/HT-5, to be published in ASME transactions, J. of Heat Transfer.
- Ingebo, R. D., (1956), "Drag coefficients for droplets and solid spheres in clouds accelerating in airstreams," National Advisory Committee for Aeronautics, Washington, D.C. Tech. Note 3762.
- Klingbiel, W. J., (1964), "Critical flow slip ratios of steam-water mixtures," Ph.D. Thesis, University of Washington, Seattle.
- Lappel, C. E., (1950), "Dust and mist collection," in Chem. Eng. Handbook, 3rd ed. (McGraw Hill Publishing Co.).
- Moody, F. J., (1965), "Maximum flow rate of a single component, two-phase mixture," Trans. ASME, Series C, 87, 134.



- Rabin, E. A., Schallenmuller, R., and R. B. Lawhead, (1960), "Displacement and shattering of propellant droplets," Final Summary Report, Rocketdyne, A Division of North American Aviation, Inc.
- Ranz, W. E., and W. R. Marshall, (1952), "Forced convection heat transfer from a single sphere," Chem. Engr. Prog., 48.
- Runge-Kutta Method as summarized, for example, in Hildebrand, F. B. (1962) Advanced Calculus for Applications (Prentice-Hall).
- Ryley, D. J., (1961), "The dispersion of water globules in steam," The Engineer.
- Smith, R. V., (1968), "Two-phase two component critical flow in a venturi," D. Phil. Thesis, Oxford, England, and Smith, R. V., Cousins, L. B., and Hewitt, G. F., (1968), "Two-phase two-component critical flow in a venturi," UKAEA Report AERE-R5736.
- Smith, R. V. (1971), "Two-phase, two-component critical flow in a venturi," ASME paper No. 71-FE-4, to be published in ASME Transactions, J. of Basic Engr.
- Wicks, Moye III and A. E. Dukler, (1966), "In Situ Measurements of drop size distribution in two-phase flow; a new method of electrically conducting liquids," 3rd Intl. Heat Transfer Conf., ASME-AIChE.
- Zivi, S. M., (1964), "Estimation of steady state steam void fraction by means of the principle of minimum energy production," Trans. ASME, Series C, 86, 247.

## 9. Nomenclature

A	=	Area, cross-sectional, flow area when single subscripted. Interface surface area when double subscripted $L^2$
c	=	specific heat $L^2/t^2T$
C	=	coefficient
D	=	tube diameter, L
$f_{fg}$	=	interface drag coefficient, dimensionless
g	=	gravitational acceleration, $L/t^2$
$h_c$	=	convection heat transfer coefficient, $M/t^3T$
k	=	thermal conductivity, $ML/t^3T$
$\ell$	=	length along venturi, L
m	=	mass flow rate per unit time, M/t
p	=	pressure, $M/Lt^2$
q	=	energy (heat) transport at interface $ML^2/t^3$
R	=	gas constant, $ML^2/t^2T$
$R_d$	=	droplet radius, L
t	=	time, t
T	=	temperature, T
u	=	velocity, $L/t$
V	=	volume, $L^3$
x	=	mixture quality, dimensionless

## Greek Letters

$\rho$  = density,  $M/L^3$

## Subscripts

c = convection (first subscript)  
= critical when subscripted to p or T  
ct = total convection  
d = droplet  
dr = drag  
e = entrained  
f = liquid  
fe = entrained liquid  
ff = liquid film  
fg = liquid-gas  
f str = equivalent liquid stream flow  
g = gas  
o = stagnation conditions (first subscript)  
p = constant pressure  
t = total  
l = venturi entrance  
= coefficient identification  
2, 3 = coefficient identification



U.S. DEPT. OF COMM. BIBLIOGRAPHIC DATA SHEET	1. PUBLICATION OR REPORT NO.  NBS-TN-608	2. Gov't Accession No.	3. Recipient's Accession No.
4. TITLE AND SUBTITLE  Steam-Water, Critical Flow in a Venturi		5. Publication Date July 1971	6. Performing Organization Code
7. AUTHOR(S) R. V. Smith		8. Performing Organization	
9. PERFORMING ORGANIZATION NAME AND ADDRESS  NATIONAL BUREAU OF STANDARDS DEPARTMENT OF COMMERCE WASHINGTON, D.C. 20234		10. Project/Task/Work Unit No. 2750478	11. Contract/Grant No.  NA
12. Sponsoring Organization Name and Address		13. Type of Report & Period Covered  NA	14. Sponsoring Agency Code
15. SUPPLEMENTARY NOTES			
<p>16. ABSTRACT (A 200-word or less factual summary of most significant information. If document includes a significant bibliography or literature survey, mention it here.)</p> <p>This paper is the second part of an analytical and experimental investigation in which the primary object was to test the hypothesis that the flow of the gas phase controls critical and near critical two-phase flow for cases where the gas and liquid flow essentially in separate streams. In the first part of the investigation, a two-component system (air-water) was used. The results presented here substantiate the hypothesis. The analytical results also indicate the use of one dimensional flow equations with reasonably accurate estimates for droplet size and for the drag and heat transfer coefficients (without consideration of mass transfer --vaporization or condensation) describe critical and near-critical flow reasonably well. This indicates that mass transfer may be a secondary effect for these flow conditions.</p>			
17. KEY WORDS (Alphabetical order, separated by semicolons) Critical flow; pressure profile; steam; venturi; water.			
18. AVAILABILITY STATEMENT  <input checked="" type="checkbox"/> UNLIMITED.  <input type="checkbox"/> FOR OFFICIAL DISTRIBUTION. DO NOT RELEASE TO NTIS.		19. SECURITY CLASS (THIS REPORT)  UNCLASSIFIED  20. SECURITY CLASS (THIS PAGE)  UNCLASSIFIED	21. NO. OF PAGES  25  22. Price  35 cents



## NBS TECHNICAL PUBLICATIONS

### PERIODICALS

**JOURNAL OF RESEARCH** reports National Bureau of Standards research and development in physics, mathematics, chemistry, and engineering. Comprehensive scientific papers give complete details of the work, including laboratory data, experimental procedures, and theoretical and mathematical analyses. Illustrated with photographs, drawings, and charts.

*Published in three sections, available separately:*

#### ● Physics and Chemistry

Papers of interest primarily to scientists working in these fields. This section covers a broad range of physical and chemical research, with major emphasis on standards of physical measurement, fundamental constants, and properties of matter. Issued six times a year. Annual subscription: Domestic, \$9.50; foreign, \$11.75\*.

#### ● Mathematical Sciences

Studies and compilations designed mainly for the mathematician and theoretical physicist. Topics in mathematical statistics, theory of experiment design, numerical analysis, theoretical physics and chemistry, logical design and programming of computers and computer systems. Short numerical tables. Issued quarterly. Annual subscription: Domestic, \$5.00; foreign, \$6.25\*.

#### ● Engineering and Instrumentation

Reporting results of interest chiefly to the engineer and the applied scientist. This section includes many of the new developments in instrumentation resulting from the Bureau's work in physical measurement, data processing, and development of test methods. It will also cover some of the work in acoustics, applied mechanics, building research, and cryogenic engineering. Issued quarterly. Annual subscription: Domestic, \$5.00; foreign, \$6.25\*.

### TECHNICAL NEWS BULLETIN

The best single source of information concerning the Bureau's research, developmental, cooperative and publication activities, this monthly publication is designed for the industry-oriented individual whose daily work involves intimate contact with science and technology—for engineers, chemists, physicists, research managers, product-development managers, and company executives. Annual subscription: Domestic, \$3.00; foreign, \$4.00\*.

\* Difference in price is due to extra cost of foreign mailing.

Order NBS publications from:

Superintendent of Documents  
Government Printing Office  
Washington, D.C. 20402

### NONPERIODICALS

**Applied Mathematics Series.** Mathematical tables, manuals, and studies.

**Building Science Series.** Research results, test methods, and performance criteria of building materials, components, systems, and structures.

**Handbooks.** Recommended codes of engineering and industrial practice (including safety codes) developed in cooperation with interested industries, professional organizations, and regulatory bodies.

**Special Publications.** Proceedings of NBS conferences, bibliographies, annual reports, wall charts, pamphlets, etc.

**Monographs.** Major contributions to the technical literature on various subjects related to the Bureau's scientific and technical activities.

**National Standard Reference Data Series.** NSRDS provides quantitative data on the physical and chemical properties of materials, compiled from the world's literature and critically evaluated.

**Product Standards.** Provide requirements for sizes, types, quality and methods for testing various industrial products. These standards are developed cooperatively with interested Government and industry groups and provide the basis for common understanding of product characteristics for both buyers and sellers. Their use is voluntary.

**Technical Notes.** This series consists of communications and reports (covering both other agency and NBS-sponsored work) of limited or transitory interest.

**Federal Information Processing Standards Publications.** This series is the official publication within the Federal Government for information on standards adopted and promulgated under the Public Law 89-306, and Bureau of the Budget Circular A-86 entitled, Standardization of Data Elements and Codes in Data Systems.

**Consumer Information Series.** Practical information, based on NBS research and experience, covering areas of interest to the consumer. Easily understandable language and illustrations provide useful background knowledge for shopping in today's technological marketplace.

**NBS Special Publication 305, Supplement 1, Publications of the NBS, 1968-1969.** When ordering, include Catalog No. C13.10:305. Price \$4.50; foreign, \$5.75.

**U.S. DEPARTMENT OF COMMERCE**  
**National Bureau of Standards**  
Washington, D.C. 20234

OFFICIAL BUSINESS

Penalty for Private Use, \$300

POSTAGE AND FEES PAID  
U.S. DEPARTMENT OF COMMERCE

